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Half Life Practice Problems Worksheet

Half-life Practice Worksheet Sodium-24 has a half-life of 15 hours. How much sodium-24 will remain in an 18.0 g sample after 60 hours? > 03 i Shr. After 42 days a 2.0 g sample of phosphorus-32 contains only 0.25 g of

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the isotope. What is the half-life of phosphorus-32? Polonium-214 has a relatively short half-life of 64 seconds. How many

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Half Life Worksheet - Extra Practice.
Fluorine-21 has a half life of

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approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute? Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days? The half-life of chromium-51 is 28 days.

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HALF-LIFE PROBLEMS Name Block 1. An isotope of cesium (cesium-137) has a

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half-life of 30 years. If 1,0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain? A We) r" 2. Actinium-226 has a half-life of 29 hours. If 100 mg of actinium-226 disintegrates over a

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half life total time of decay number of half lives--= 3. Calculate how many half-lives have passed during the decay of the 100.0 g sample. fraction of sample remaining = final mass of sample initial mass of sample g g == 25 0 100 0 1 4..

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after one half-life = 1 2; after two half-lives = 1 2 1 2 1 4 \times = of sample Two half-lives have passed. 4. Solve for the half-life.

Half-Life - Southside High School

Practice Problems. Problem 1 : The half-life of carbon-14 is approximately 6000 years. How much of 800 g of this

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substance will remain after 30,000 years? Solution : Half-Life Decay Formula : $A = P(1/2)^{t/d}$. Substitute. $P = 800$. $t = 30000$. $d = 6000$. Then, $A = 800 (1/2)^{30000/6000}$. $A = 800 (1/2)^5$. $A = 800 (0.5)^5$. $A = 800(0.03125)$ $A = 25$

**Half Life Decay Formula -
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Problem #3: Os-182 has a half-life of 21.5 hours. How many grams of a 10.0 gram sample would have decayed after exactly three half-lives? Solution: $(1/2)^3 = 0.125$ (the amount remaining after 3 half-lives) $10.0 \text{ g} \times 0.125 = 1.25 \text{ g}$ remain $10.0 \text{ g} - 1.25 \text{ g} = 8.75 \text{ g}$ have decayed Note that the length of the half-life played no role in this calculation.

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ChemTeam: Half-Life Problems #1 - 10

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Word Problems: Interest,

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Half-Life Practice Problems . 1.) What is the half-life of a 100.0 g sample of nitrogen-16 that decays to 12.5 grams in 21.6 seconds? 2.) All isotopes of

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technetium are radioactive, but they have widely varying half-lives. If an 800.0 gram sample of technetium-99 decays to 100.0 g of technetium-99 in 639,000 years, what is its half-life? 3.)

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themed worksheet is a two-page worksheet that helps students practice setting up simple half-life decay tables. The students have to figure out how much of a substance will be left after a given amount of

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Sodium-24 has a half-life of 15 hours. How much sodium-24 will remain in an 18.0 g sample after 60 hours? 30

2. After 42 days a 2.0 g sample of thorium-232 contains one-eighth of the original isotope. What is the half-life of

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phosphorus-32? 3. Polonium-214 has a relatively short half-life of 164 seconds. How many

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